



Photocatalysis of the Aminophosphonate EDTMP

Insights into Kinetics and Degradation Mechanisms of
EDTMP via Doped TiO_2 Photocatalysts

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- **photochemical**
Degradation of EDTMP
- **additives** H_2O_2 ,
Metals...
- few publications on
industrial
phosphonates

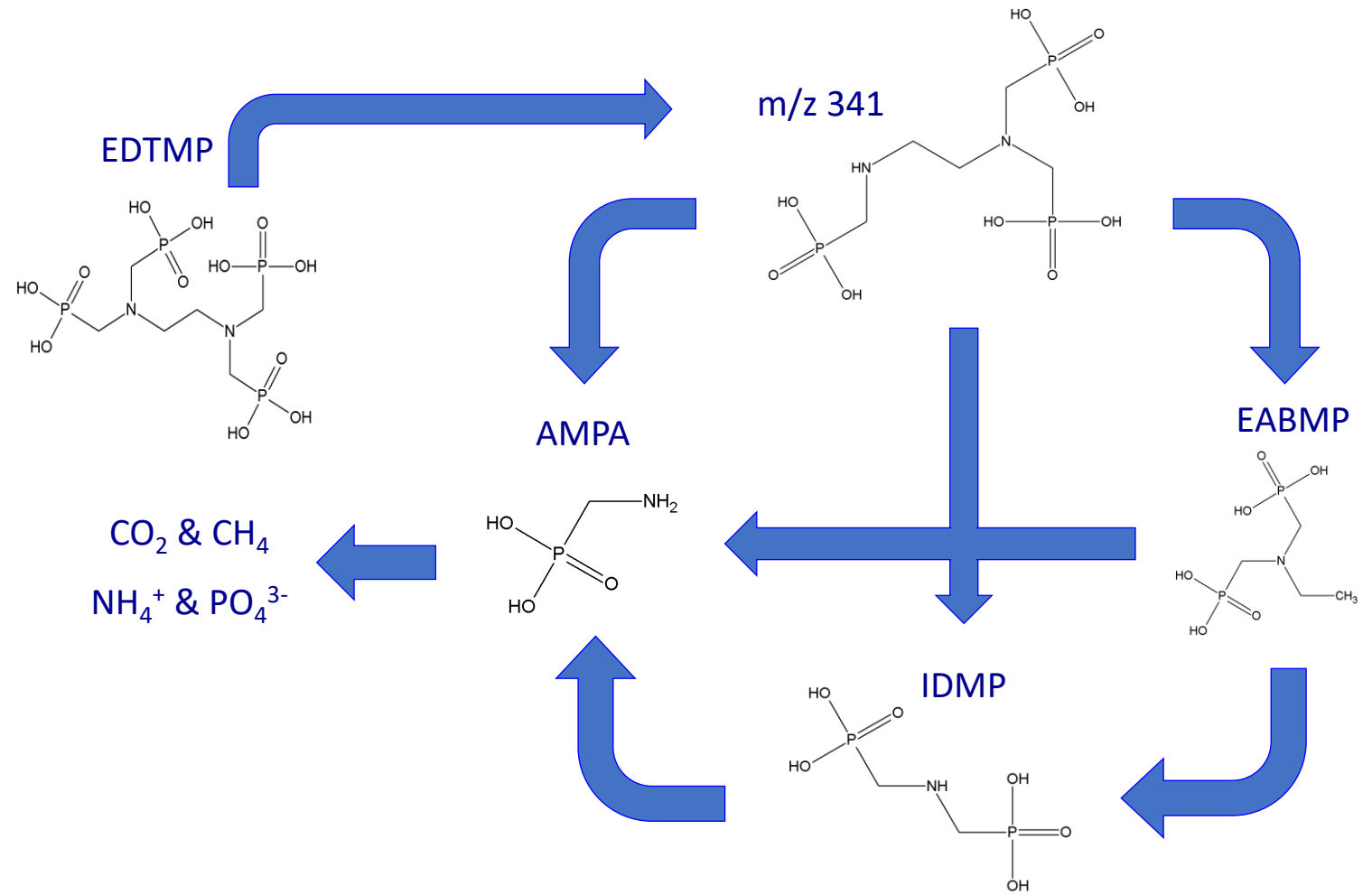


Fig. 1. Photochemical degradation pathway of EDTMP.

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Degradation of EDTMP
 - **additives** H_2O_2 ,
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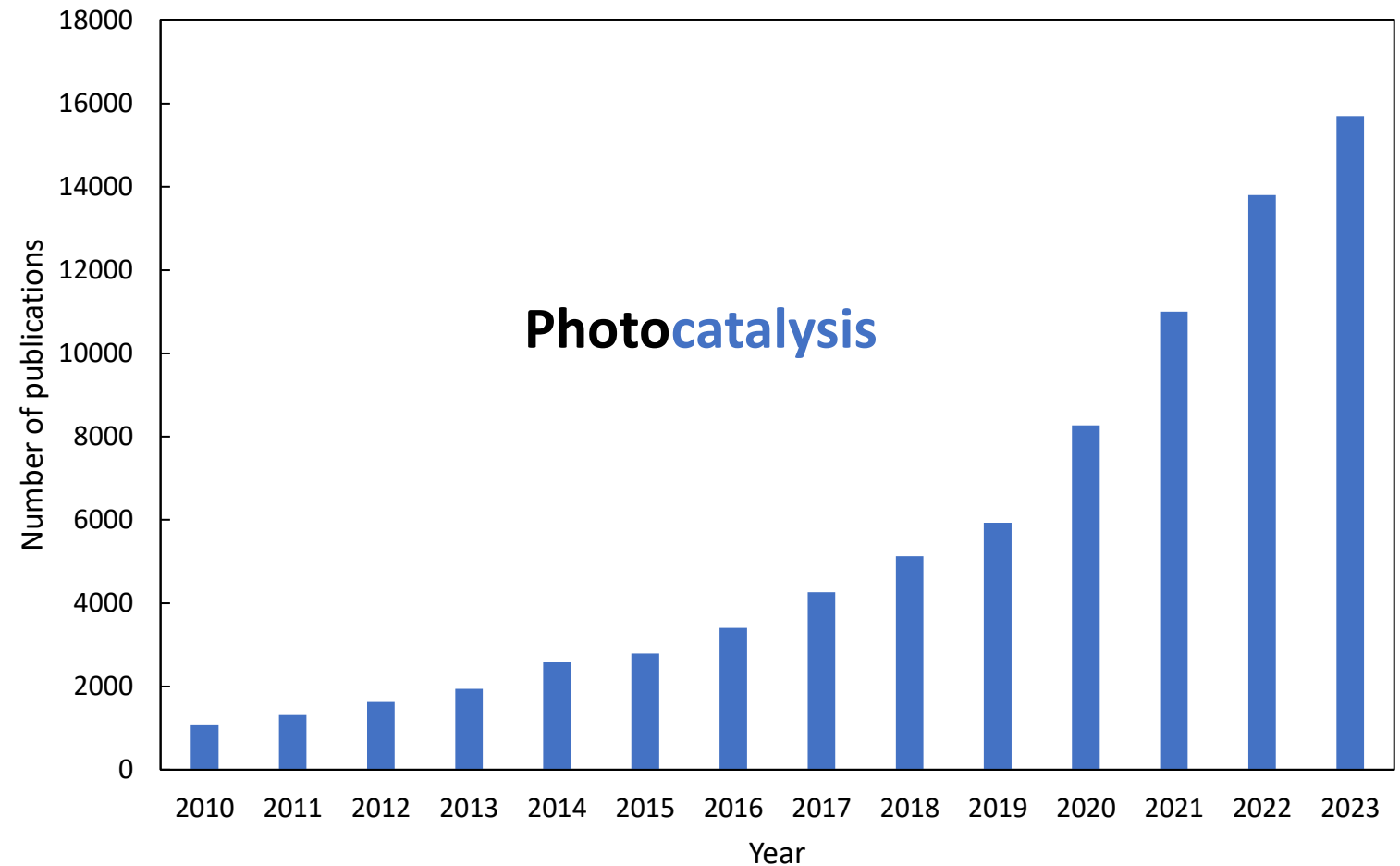
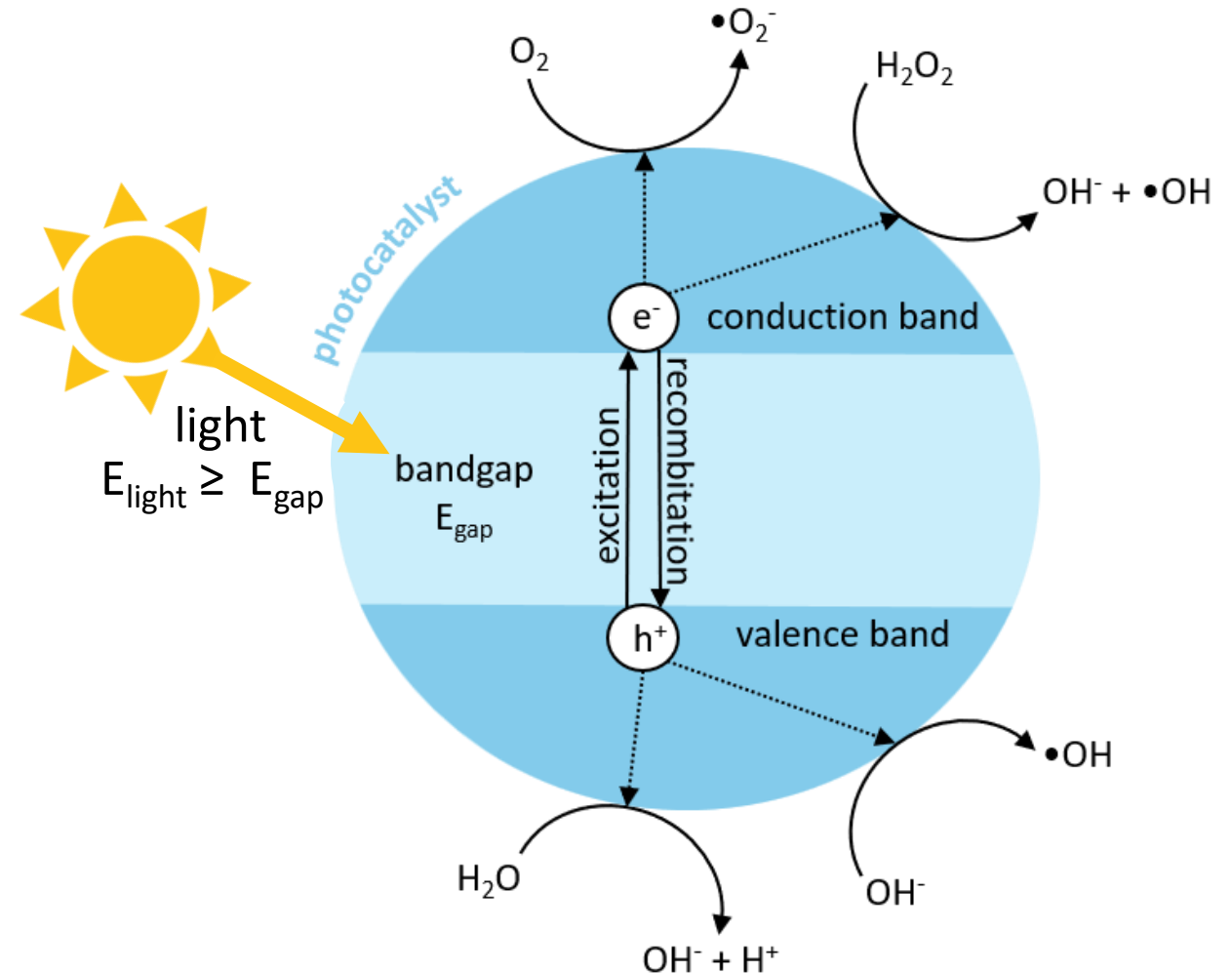


Fig. 2. Publications per year regarding photocatalysis of persistent chemicals.

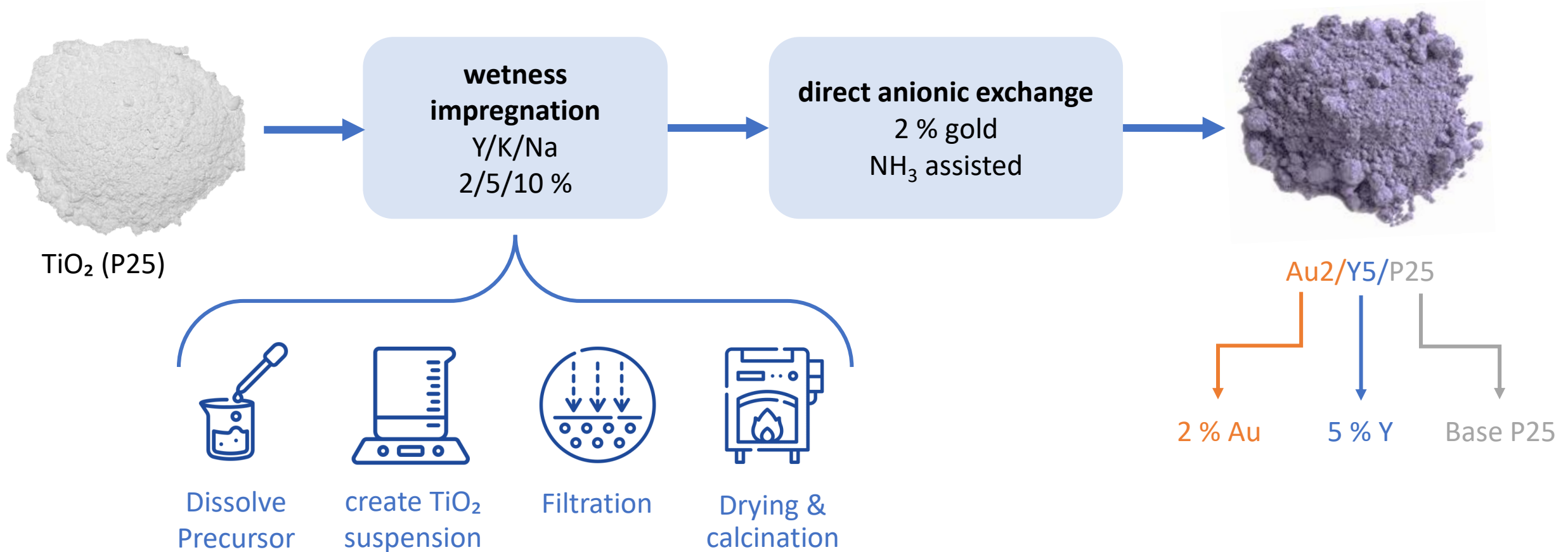
<https://doi.org/10.1016/j.jphotochem.2019.112192>

<https://scholar.google.de/scholar?q=photocatalysis+persistent+chemicals>

1. **direct:** specific target degradation
 - Target needs physical contact to catalyst surface
 - e^-/h^+ transfer to target is possible
2. **indirect:** ROS attack target
 - More effective for degradation of organic pollutants*



Synthesis of the TiO_2 -supported NP photocatalysts



- **confirmed:** rutile & anatase phase

Au-presence

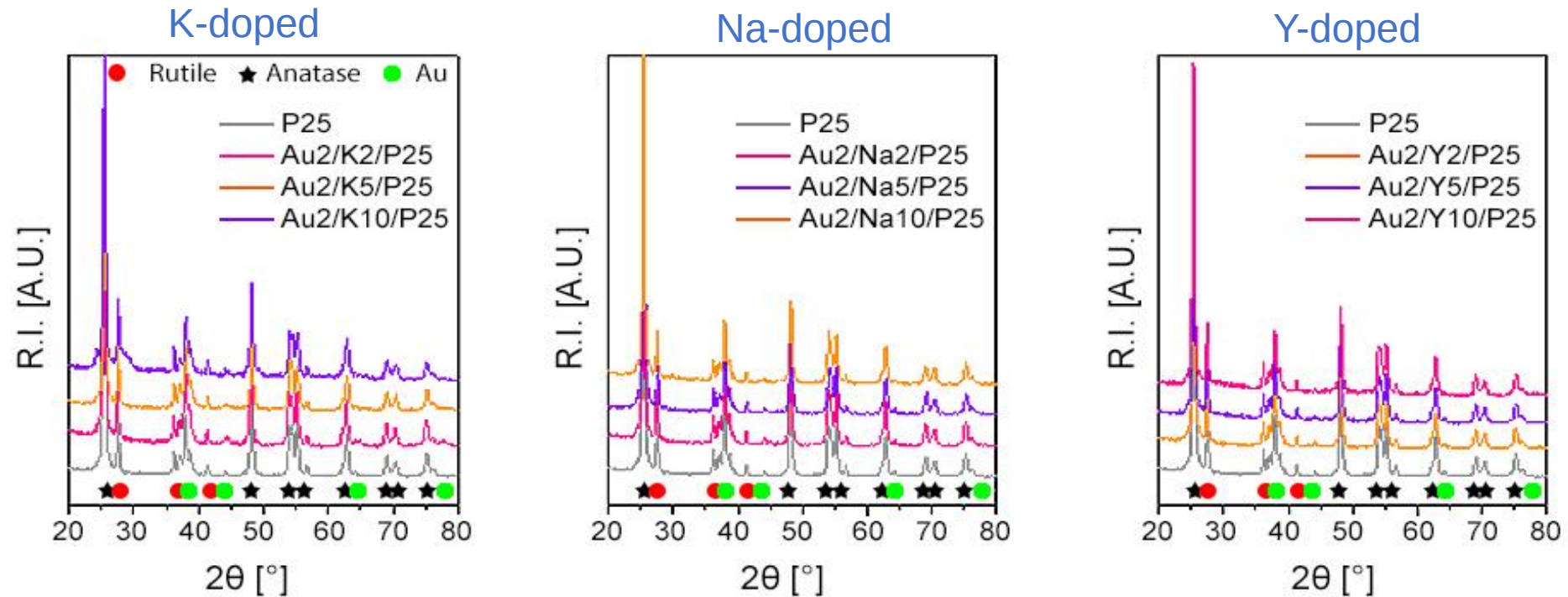


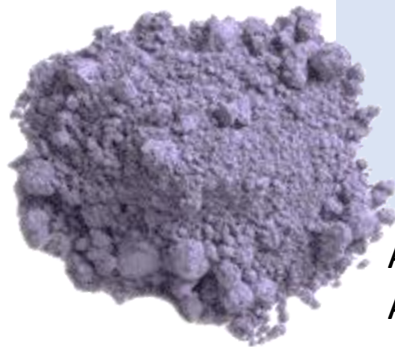
Fig. 3. XRD diffractograms obtained for the Au catalyst series.

Characterisation of NPs: UV-VIS

- **Au:** 550 nm peak → purple appearance
- **K & Y:** increase of absorption



Depends on dopant
amount!



Au2/K2/P25
Au2/K10/P25

Au2/Y2/P25
Au2/Y5/P25

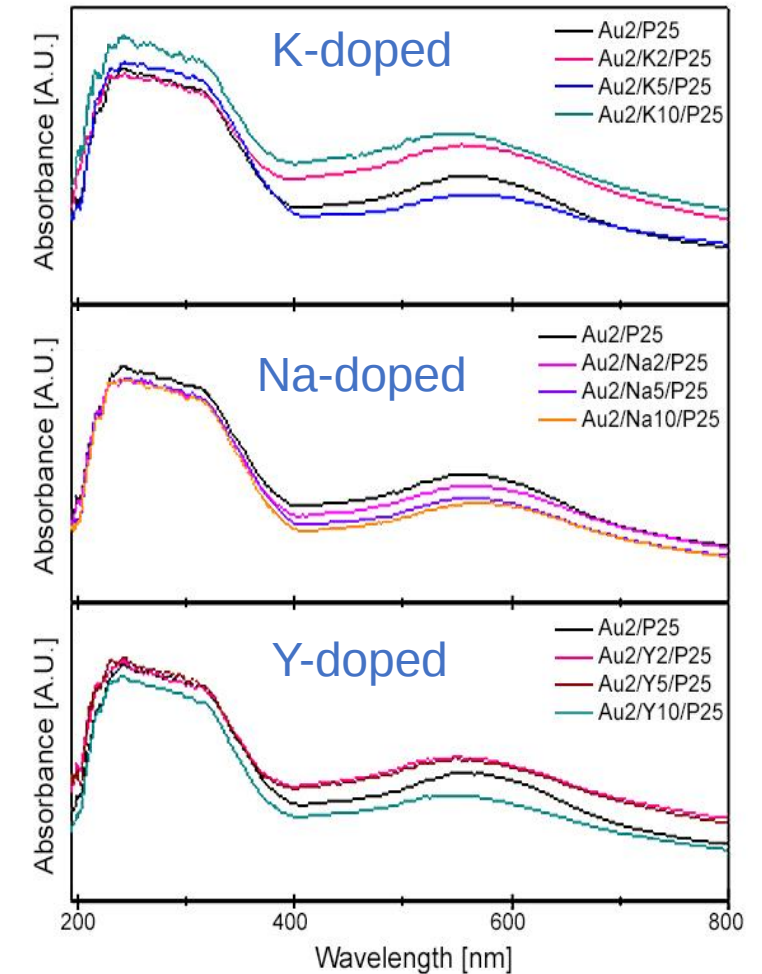


Fig. 4. UV-Vis spectra obtained for the Au catalysts series.

Characterisation of NPs: TEM

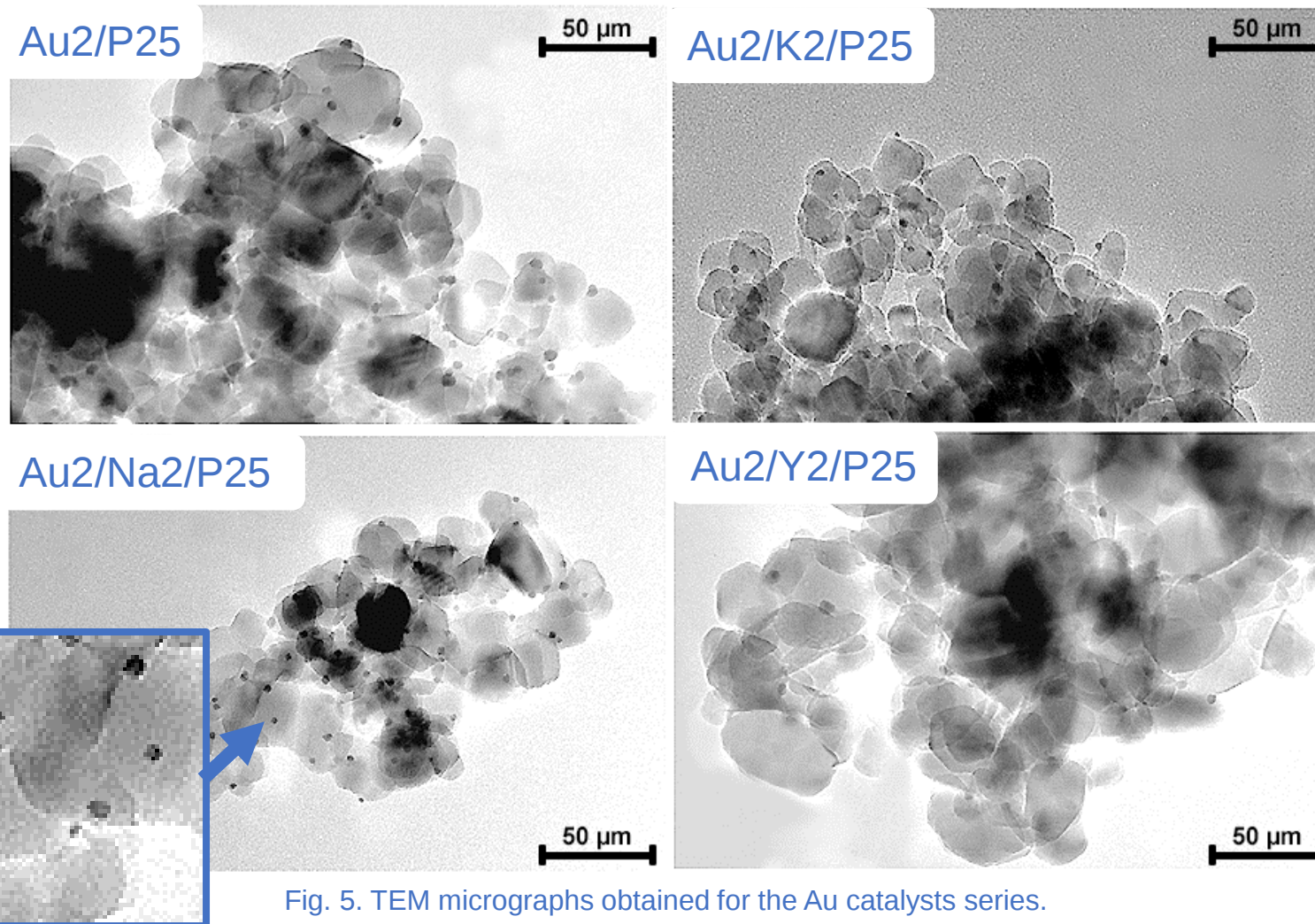


Fig. 5. TEM micrographs obtained for the Au catalysts series.

- spherical anatase
- angular rutile
- round small gold NPs

- **Y-doping:** smallest particles
- **Na-doping:** biggest particles

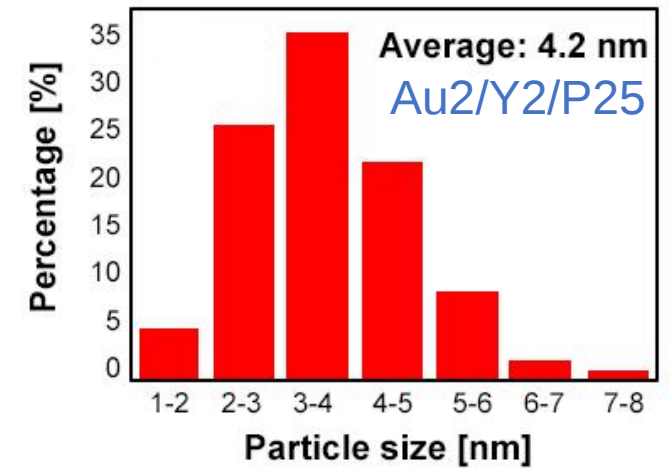
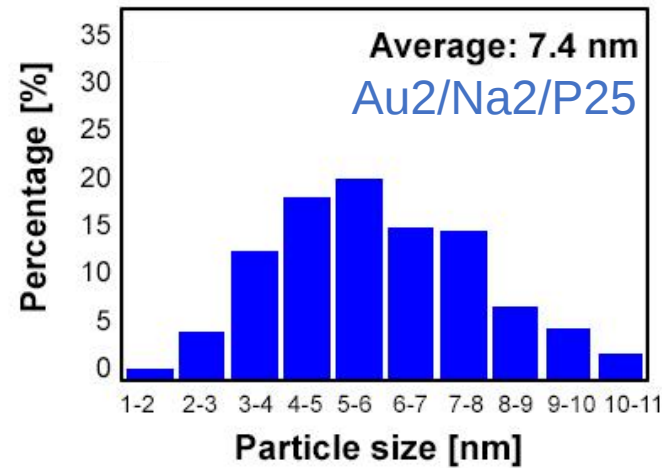
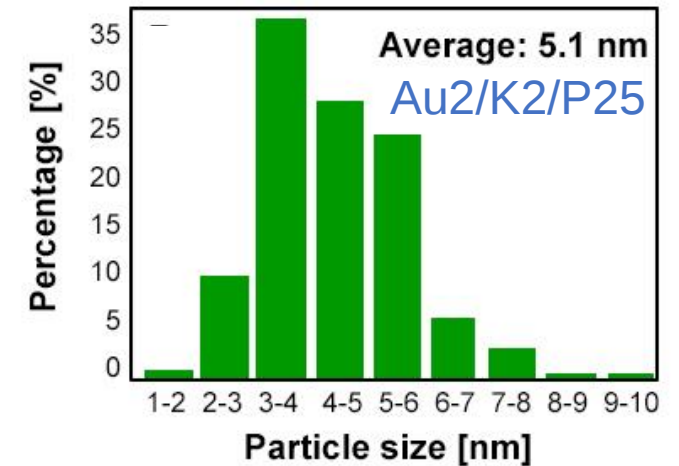
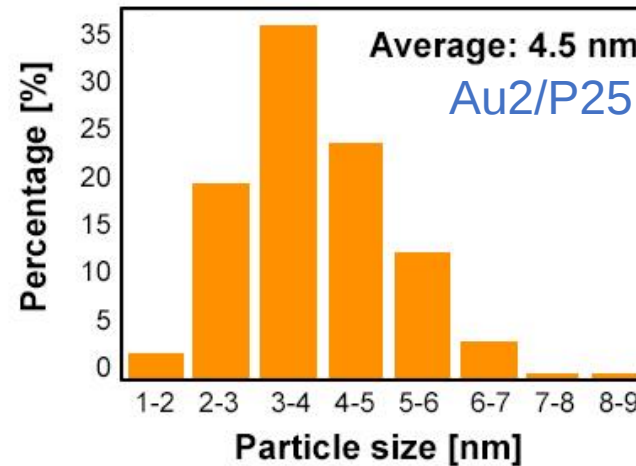


Fig. 6. Histograms of particle size distribution for the Au catalysts series.

Pre-screening photocatalysis experiments

- lowest performance: acidic pH

- best performance: pH 7

Au2/Y2/P25

Au2/Y5/P25

Au2/K2/P25

Au2/K10/P25

pH 10

Na-doped

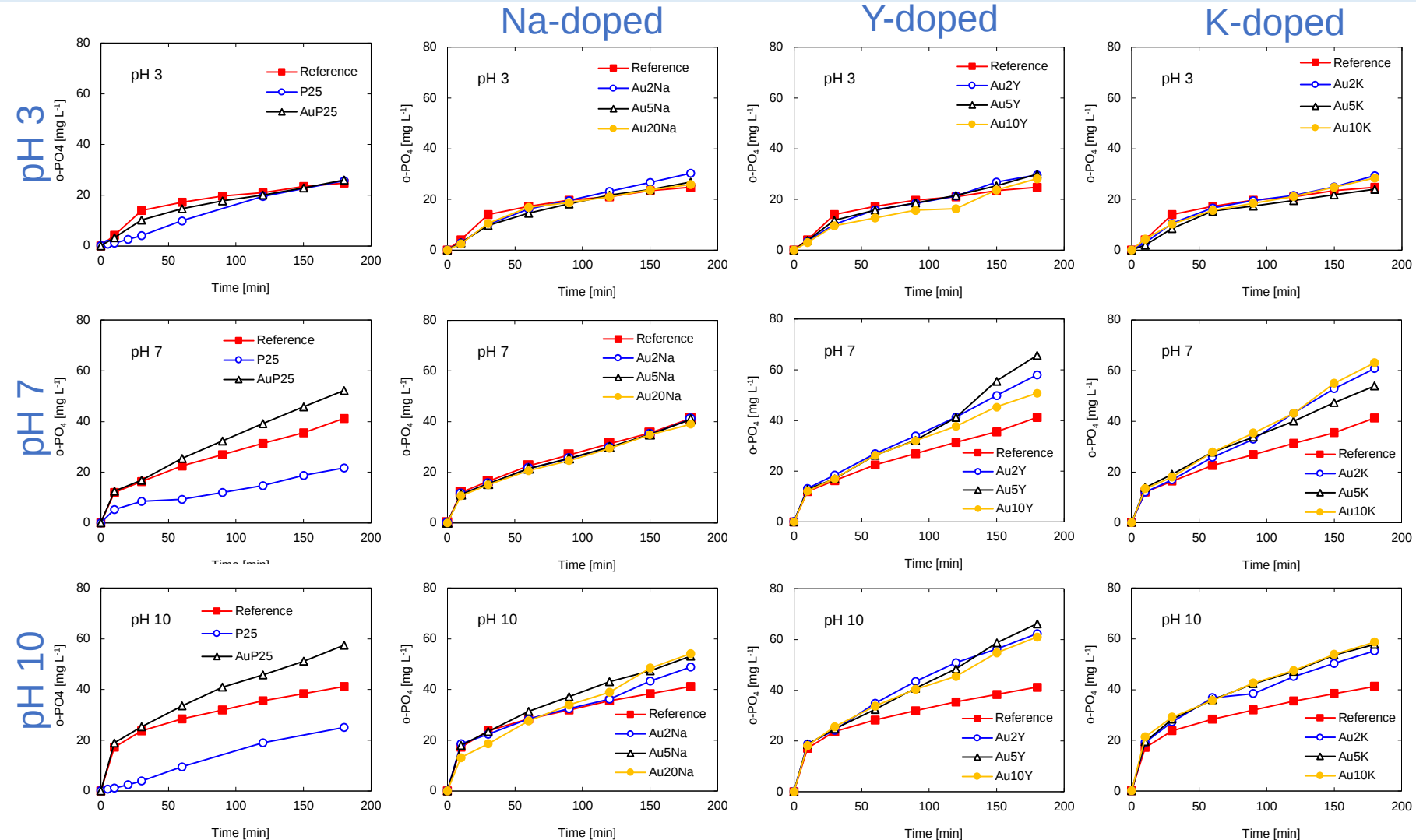
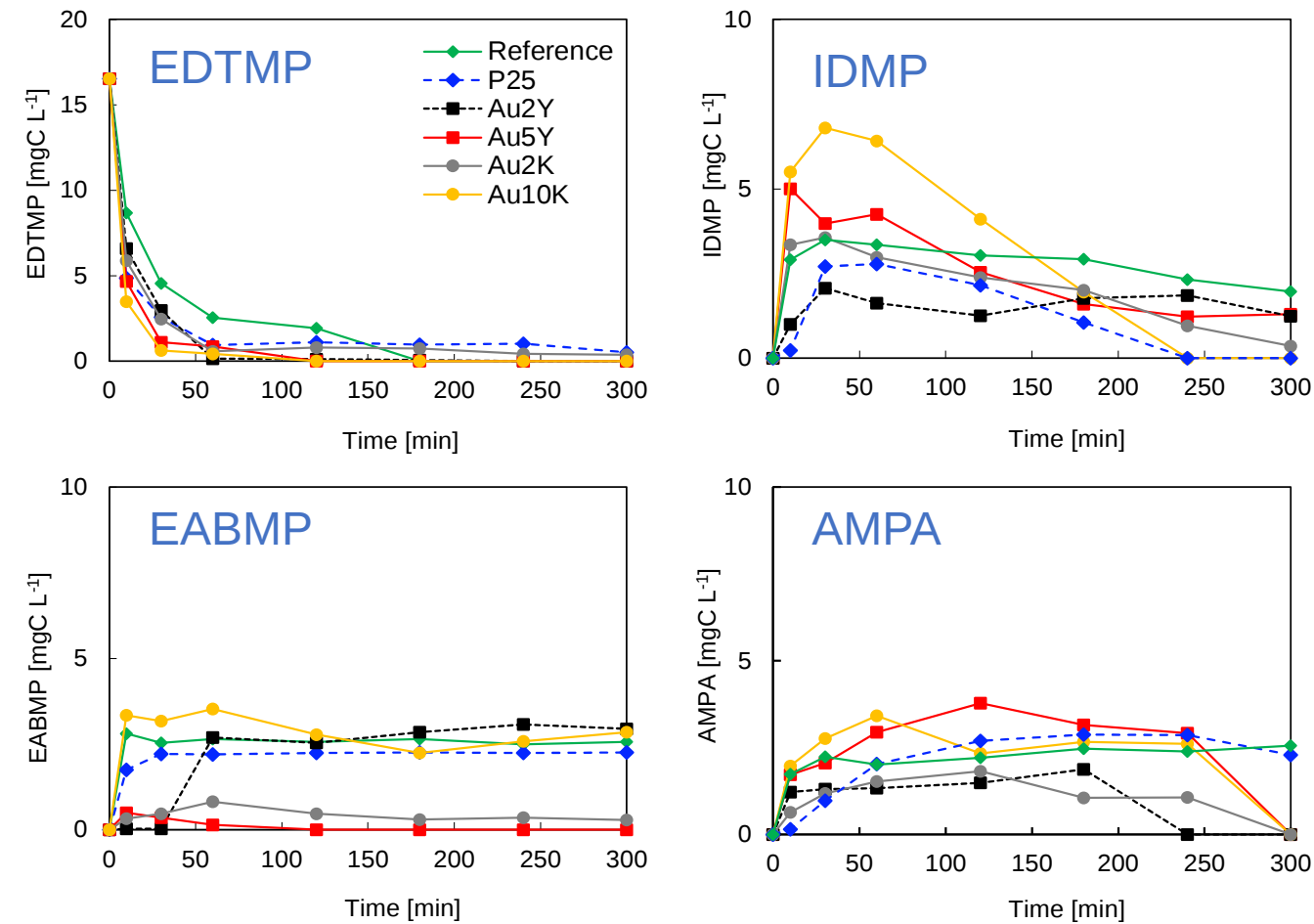


Fig. 7. $o\text{-PO}_4^{3-}$ Release of pre-screening photocatalysis (100 mg L⁻¹ EDTMP, pH 3;7;10).

Performance of selected NPs



Tab. 1. o-PO_4^{3-} release and mass balance.

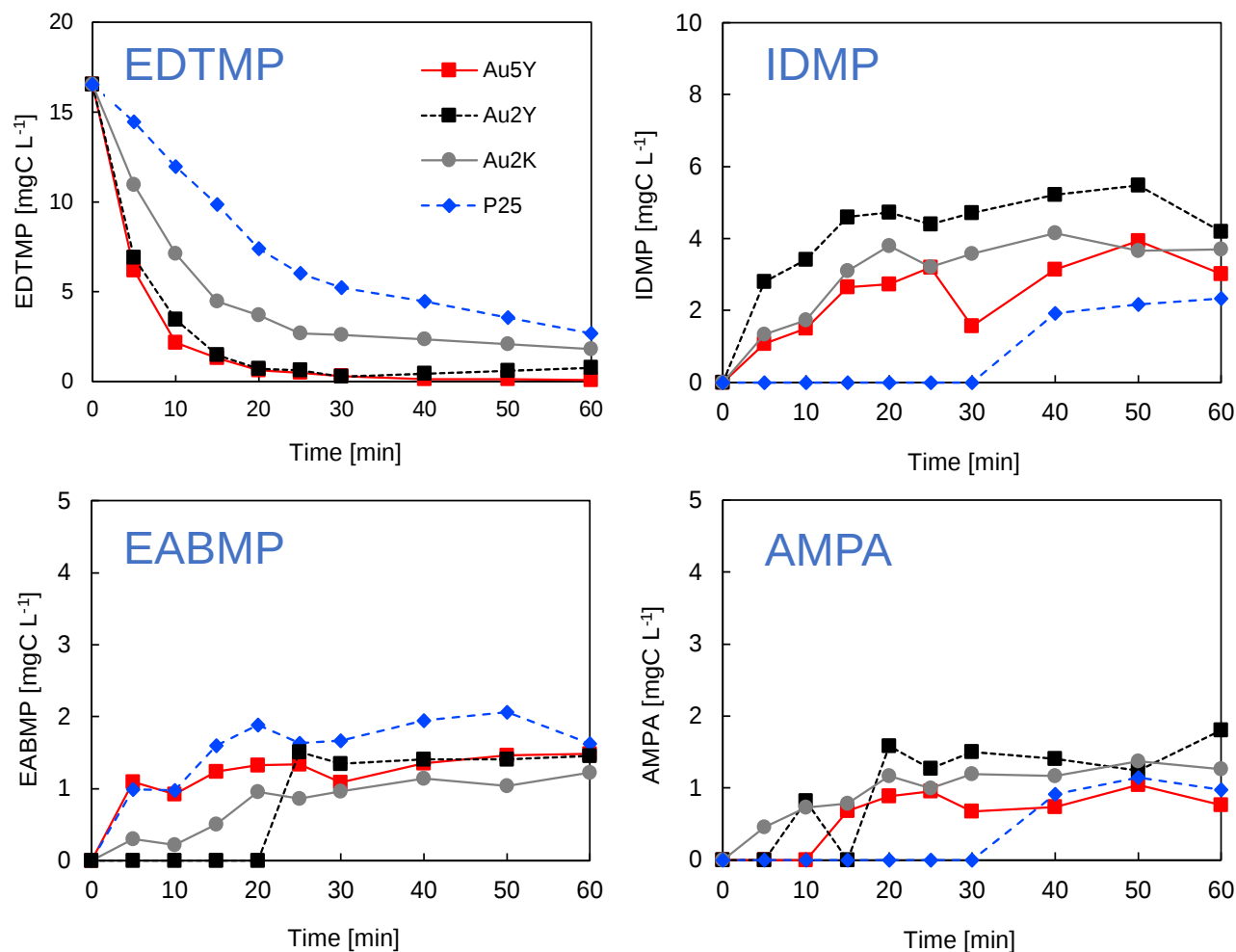
[%]	o-PO_4^{3-} release	P gap [%]	C gap [%]	N gap [%]
Reference	47	0	54	21
Au2/Y2/P25	88	0	74	75
Au2/Y5/P25	92	0	92	88
Au2/K2/P25	79	15	96	95
Au2/K10/P25	88	0	82	87
P25	65	0	69	45

- rapid degradation of EDTMP in 60 min
- doubled o-PO_4^{3-} release Au2/Y5/P25

LC-MS

Fig. 8. Long-term photocatalysis (100 mg L^{-1} EDTMP, pH 7).

Influence of selected NPs on the kinetics



Tab. 2. Degradation rate constants and half-life of EDTMP.

Catalyst	$k_{uv} [s^{-1}]$	Half-life [min]	RSS*
Au2/Y2/P25	2.9 E^{-3}	3.9	8.4
Au2/Y5/P25	3.9 E^{-3}	2.9	54.4
Au2/K2/P25	1.9 E^{-3}	6.0	60.7
P25	9.8 E^{-4}	11.7	54.1

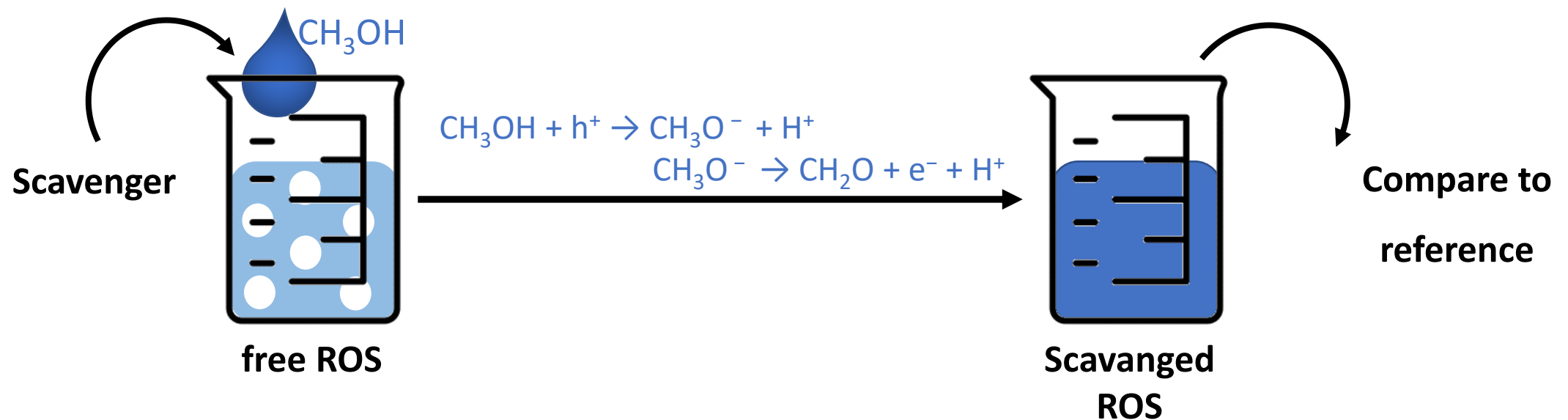
- highest k_{uv} & lowest $t_{1/2}$: Au2/Y5/P25

LC-MS

Fig. 9. 60-min-photocatalysis (100 mg L^{-1} EDTMP, pH 7).

*RSS: Residual sum of squares

- **aim:** investigate reactive species responsible for target degradation
- **method:**



Determination of ROS by scavengers

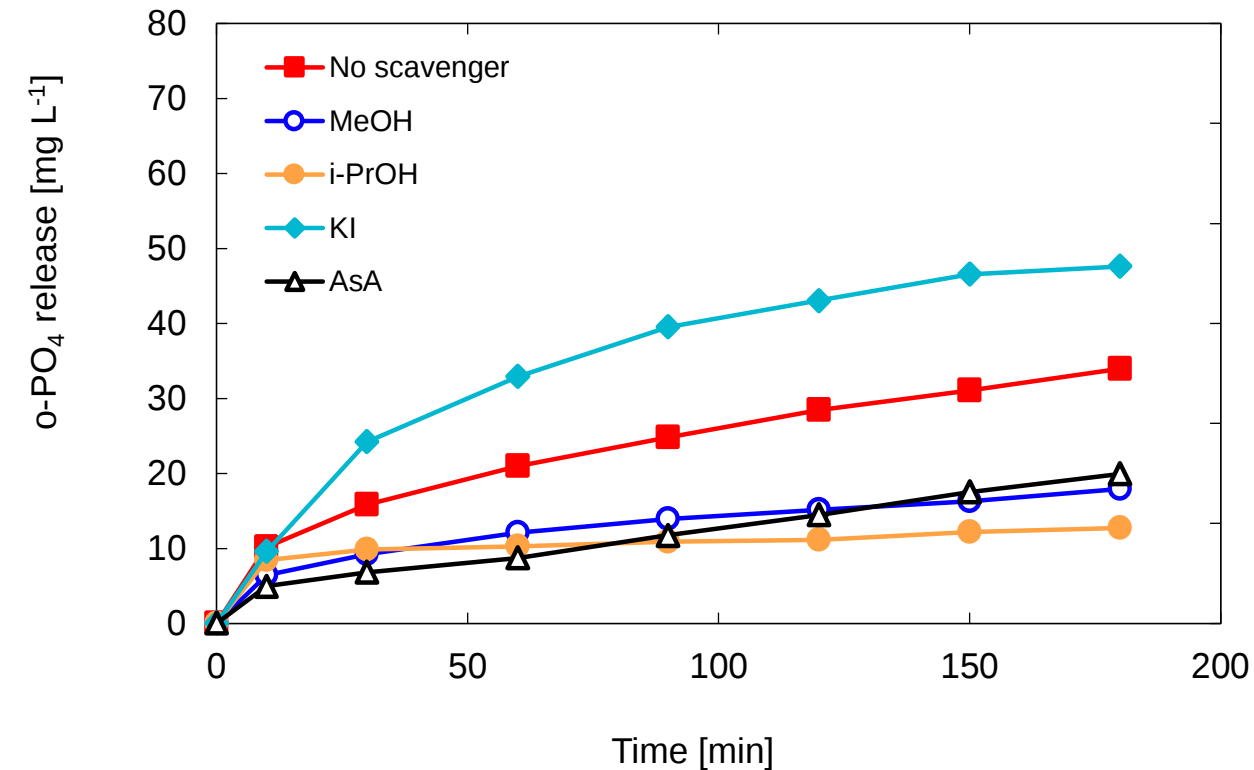


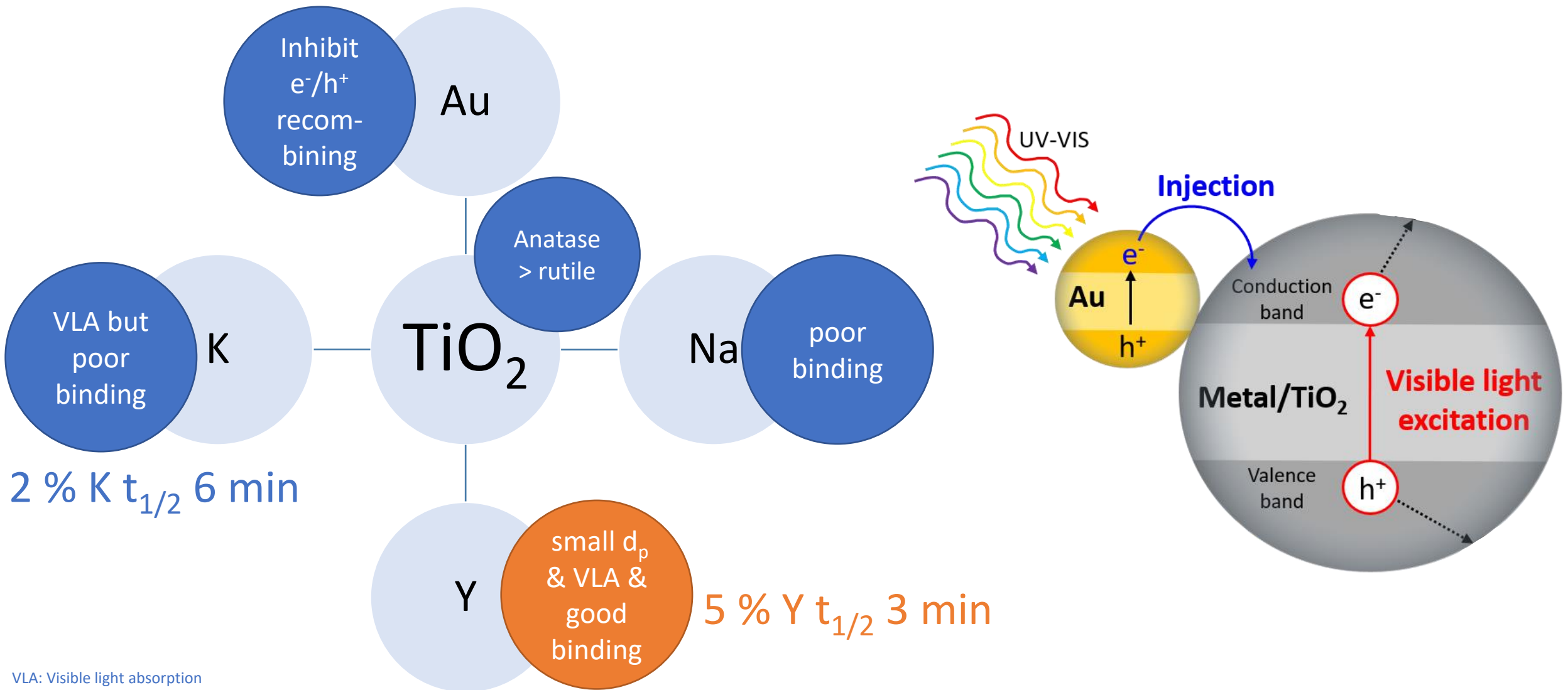
Fig. 10. Scavenger tests with 100 mg L⁻¹ EDTMP and 100 mg L⁻¹ Au₂/Y₅/P₂₅, pH 7.

Tab. 2. P release of EDTMP during photocatalysis (Au₂/Y₅/P₂₅, pH 7, scavenger).

Condition	Scavenged ROS	Scavenged medium	P release	
			o-PO ₄ ³⁻ [mg L ⁻¹]	o-PO ₄ ³⁻ [%]
Without	-	-	33	38
MeOH	h ⁺ & •OH	Bulk and surface	19	21
i-PrOH	•OH	Bulk	12	14
KI	•OH & h ⁺	Surface	47	54
AsA	•O ₂ ⁻	Bulk	14	17

- **primary ROS: •OH (bulk) & h⁺ (surface)**

Influence of the dopants



VLA: Visible light absorption

- degradation mechanism **similar** to photochemical processes
- **4 fold enhanced** kinetics

